Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

## Shu-Sheng Zhang,* Yong-Hong

Wen, Juan Liang and Xue-Mei Li
College of Chemistry and Molecular
Engineering, Qingdao University of Science and Technology, 266042 Qingdao, Shandong,
People's Republic of China
Correspondence e-mail:
zhangshush@public.qd.sd.cn

## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.006 \AA$
$R$ factor $=0.085$
$w R$ factor $=0.257$
Data-to-parameter ratio $=16.1$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
(C) 2004 International Union of Crystallography Printed in Great Britain - all rights reserved

## $N, N^{\prime}$-Bis(p-tolyl)-2,2'-(m-phenylenedioxy)diacetamide

In the title compound, $\mathrm{C}_{24} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{4}$, the molecules are linked into a three-dimensional network by intermolecular N $\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds. The packing is further stabilized by $\mathrm{C}-\mathrm{H} \cdots \pi$ interactions.

## Comment

We have reported a novel amide-type acyclic polyether with 1,4 -dihydroxybenzene as a skeleton, namely $N, N^{\prime}$-bis( $p$-tolyl)-2,2'-(p-phenylenedioxy)diacetamide, (II) (Wen et al., 2004), which is a possible active material for ion-selective electrodes or a fine extractor of metal ions. In our ongoing study of the effect of the skeleton of the acyclic polyether on the coordination selectivity and extractability of metal ions, we have carried out a similar reaction for $N$-( $p$-tolyl)-2-chloroacetamide with 1,3-dihydroxybenzene, and obtained the title compound, (I) (Fig. 1).

(I)

The bond lengths (Table 1) and angles in (I) are comparable with those in (II). The molecule in (II) is essentially planar, while the molecule in (I) has a wing-like conformation. In (I), the dihedral angles between the central benzene ring ( $\mathrm{C} 10-$ C 15 ) and the two outer benzene rings ( $\mathrm{C} 2-\mathrm{C} 7$ and $\mathrm{C} 18-\mathrm{C} 23$ ) are 28.2 (2) and $86.9(3)^{\circ}$, respectively. Both the outer benzene rings are involved in intramolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions, with the two carbonyl O atoms as acceptors (Table 2). Molecules of (I) are linked into a three-dimensional network via intermolecular $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds (Table 2). The packing is further stabilized by $\mathrm{C}-\mathrm{H} \cdots \pi$ interactions (Table 2).

Received 14 July 2004 Accepted 19 July 2004 Online 24 July 2004


Figure 1
The structure of (I), showing $50 \%$ probability displacement ellipsoids and the atom-numbering scheme.
amine, according to the literature method of Li (2001). To a solution of $N$-(p-tolyl)-2-chloroacetamide ( $1.84 \mathrm{~g}, 10 \mathrm{mmol}$ ) in acetone ( 35 ml ) was added 1,3-dihydroxybenzene ( $0.55 \mathrm{~g}, 5.0 \mathrm{mmol}$ ), $\mathrm{K}_{2} \mathrm{CO}_{3}$ $(1.52 \mathrm{~g}, 11 \mathrm{mmol})$ and $\mathrm{KI}(0.5 \mathrm{~g})$, and the mixture stirred at 328 K for 5 h . After cooling to room temperature, the mixture was washed three times with water and then filtered. The filtered solid was recrystallized from alcohol and water. The title compound was obtained after drying the yellow powder at room temperature for 48 h . Light-brown single crystals suitable for X-ray diffraction analysis were obtained by slow evaporation of a DMF-ethanol (1:20 $v / v)$ solution over a period of one month.

## Crystal data

$\mathrm{C}_{24} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{4}$
$M_{r}=404.45$
Orthorhombic, Pbca
$a=18.5029(11) \AA$
$b=10.6914(7) \AA$
$c=21.3741(13) \AA$
$V=4228.3(5) \AA^{3}$
$Z=8$
$D_{x}=1.271 \mathrm{Mg} \mathrm{m}^{-3}$

Mo $K \alpha$ radiation
Cell parameters from 3272
reflections
$\theta=2.9-26.0^{\circ}$
$\mu=0.09 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Block, brown
$0.30 \times 0.26 \times 0.24 \mathrm{~mm}$

## Data collection

Bruker SMART 1000 CCD areadetector diffractometer

## $\omega$ scans

Absorption correction: multi-scan (SADABS; Sheldrick, 1996) $T_{\text {min }}=0.974, T_{\max }=0.979$
21758 measured reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.085$
$w R\left(F^{2}\right)=0.257$
$S=1.02$
4172 reflections
259 parameters
H -atom parameters constrained

Table 1
Selected distances ( A ).

| O1-C8 | $1.225(4)$ | O4-C17 | $1.229(4)$ |
| :--- | :--- | :--- | :--- |
| O2-C10 | $1.368(4)$ | N1-C8 | $1.347(4)$ |
| O2-C9 | $1.413(4)$ | N1-C5 | $1.411(4)$ |
| O3-C14 | $1.375(4)$ | N2-C17 | $1.330(4)$ |
| O3-C16 | $1.413(4)$ | N2-C18 | $1.410(5)$ |

Table 2
Hydrogen-bonding geometry $\left(\AA,^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{N} 1-\mathrm{H} 1 A \cdots \mathrm{O} 4^{\mathrm{i}}$ | 0.86 | 2.09 | 2.946 (4) | 176 |
| $\mathrm{N} 2-\mathrm{H} 2 A \cdots \mathrm{O} 1^{\text {ii }}$ | 0.86 | 2.06 | 2.912 (4) | 172 |
| C6-H6A $\cdots$ O1 | 0.93 | 2.24 | 2.844 (5) | 122 |
| C19-H19A $\cdots$ O 4 | 0.93 | 2.41 | 2.926 (7) | 115 |
| $\mathrm{C} 9-\mathrm{H} 9 A \cdots \mathrm{Cg} 1^{\text {iii }}$ | 0.97 | 2.80 | 3.716 | 158 |
| $\mathrm{C} 9-\mathrm{H} 9 \mathrm{~B} \cdots \mathrm{Cg} 2^{\text {i }}$ | 0.97 | 2.88 | 3.639 | 135 |

Symmetry codes: (i) $-x, 1-y,-z$; (ii) $-\frac{1}{2}-x, y-\frac{1}{2}, z$; (iii) $-x, 2-y,-z . C g 1$ and $C g 2$ denote the centroids of the $\mathrm{C} 2-\mathrm{C} 7$ and $\mathrm{C} 10-\mathrm{C} 15$ benzene rings

All H atoms were positioned geometrically, with $\mathrm{C}-\mathrm{H}=0.93-$ $0.96 \AA$ and $\mathrm{N}-\mathrm{H}=0.86 \AA$, and treated as riding, with $U_{\text {iso }}(\mathrm{H})=1.2$ or 1.5 times $U_{\text {eq }}$ of the carrier atoms.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 1990).

This project was supported by the National Natural Science Foundation of China (No. 20275020), the State Key Laboratory Foundation of Electroanalytical Chemistry of the Changchun Institute of Applied Chemistry (No. SKLEAC 2004-8) and the Outstanding Adult-Young Scientific Research Encouraging Foundation of Shandong Province (No. 03BS081).

## References

Li, Z. G. (2001). Preparation of Organic Intermediates, pp. 103-107. Beijing: Chemical Industry Press.
Nardelli, M. (1995). J. Appl. Cryst. 28, 659.
Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
Sheldrick, G. M. (1997). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
Siemens (1996). SMART and SAINT. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
Spek, A. L. (1990). Acta Cryst. A46, C-34.
Wen, Y.-H., Zhang, S.-S., Liang, J. \& Li, X.-M. (2004). Acta Cryst. E60, o1198o1199.

